



# A Polymer Flooding Mechanism for Mature Oil Fields: Laboratory Measurements and Field Results Interpretation



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Submission: January 18, 2018; Published: February 06, 2018

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## Abstract

During propagation in mature fields, the polymer dispersion can undergo different concentration regimes. This change of regime in the porous medium can promote a significant reduction of the water effective permeability, making this the main mechanism of the polymer flooding process in mature fields.

**Keywords:** Enhanced oil recovery; Polymer flooding, Particle size, Rheology; Permeability

## Introduction

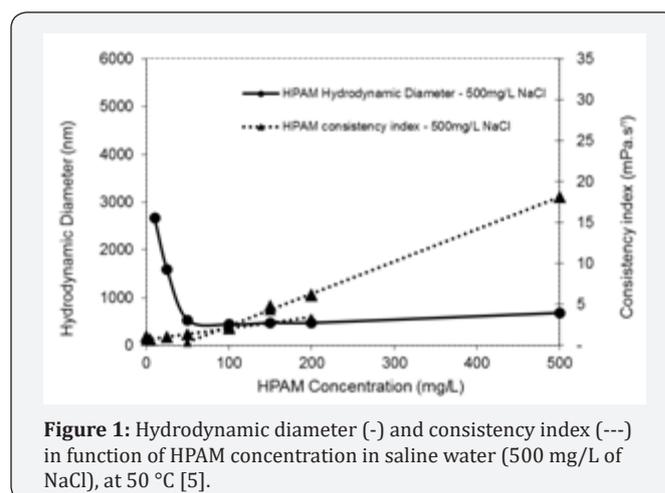
Polymer flooding is an enhanced oil recovery that consists in addition of polymers with high thickening potential in the injection water [1-3]. The polymer increases the aqueous phase viscosity promoting the mobility ratio reduction and increasing the sweep efficiency. Additionally, depending on the polymer type, the effective permeability to water in the zones swept by the polymer can be reduced [4]. These are the fundamental mechanisms widely accepted.

During propagation in mature fields, the polymer dispersion preferably flows into the more permeable and washed zones, which can lead to greatly reduced concentration and viscosity of polymer dispersion. Depending on the dilution of polymer dispersion in these zones, the polymer dispersion may undergo different concentration regimes. A research based on the evaluation of hydrodynamic diameter ( $D_h$ ), rheological behavior and flow tests on porous media showed that the static and thermodynamic properties of Partially Hydrolyzed Polyacrylamide (HPAM) are markedly impacted by these concentration regimes [5].

## Particle Size and Rheological Evaluations

The HPAM in this study was 5-6.106 nominal molar mass and 30% hydrolysis degree. The diffusion coefficient was measured by dynamic light scattering (DLS) and introduced in the

Stokes-Einstein equation to convert to hydrodynamic size. The consistency index ( $K$ ), which represents the thickening potential, was obtained with Power-law model adjustment.



**Figure 1:** Hydrodynamic diameter (-) and consistency index (---) in function of HPAM concentration in saline water (500 mg/L of NaCl), at 50 °C [5].

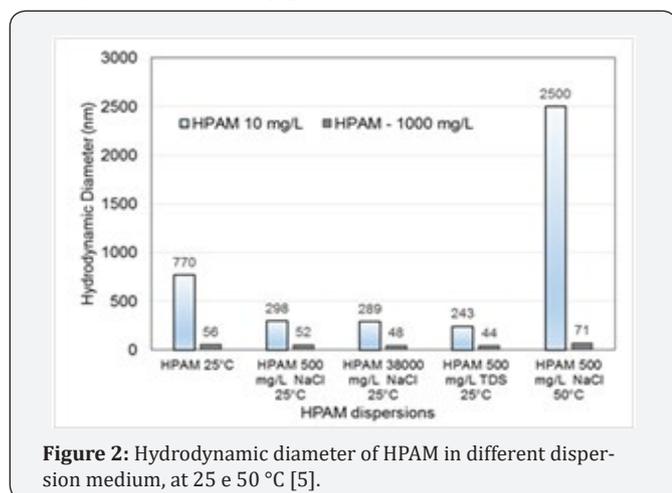
Concentration influence. The evaluation of the  $D_h$  and consistency index of the dispersions of HPAM in saline) solution 500mg/L (NaCl) were analyzed in the range of 10 to 1000mg/L and at temperature of 50°C. The  $D_h$  versus concentration curves presented in Figure 1 show an inflection point at approximately 50mg /L and the consistency index versus concentration at 100 mg/L. These values of overlapping concentration (\*) marks the

separation between the diluted and semi-diluted regime. The difference of values of  $\eta^*$  has been reported by other authors and it has been attributed to the differences in shear applied by the two techniques, since an increase in shear causes polymer chain elongation, hindering interpenetration and increasing the value of  $\eta^*$  detected [6].

The rheological analysis shows that the thickening potential ( $K$ ) of the HPAM increases markedly at concentrations greater than overlapping ( $\eta^*$ ), while the hydrodynamic diameter tends to its lower value (unperturbed value).

The  $D_h$  data of the analyzed HPAM dispersions show great variation as a function of the concentration. The  $D_h$  increase was greater at lower concentrations (dilute regime), probably because the smaller the number of dispersed macromolecules:

- (i) the weaker the repulsion between the macromolecules is; and
- (ii) the greater the hydration is, leading to a larger hydrodynamic diameter [7].



**Figure 2:** Hydrodynamic diameter of HPAM in different dispersion medium, at 25 and 50 °C [5].

The influence of salinity and temperature on the HPAM hydrodynamic diameter is shown at Figure 2. The polymer concentrations were chosen to represent the dilute and concentrated regimes. Besides this, a softened injection water, treated to remove divalent cations (500mg/L, Dissolved solids total) was also analyzed [8].

### Concentration influence

The Figure 2 shows the dispersions of HPAM at concentration of 10mg/L (dilute regime) had larger  $D_h$  than those at concentration of 1,000mg/L (concentrated regime) under all the conditions analyzed. The increase of HPAM concentration from 10 to 1000mg/L decreases the  $D_h$  from 770nm to 56nm, at 25 °C [9].

### Salinity influence

For both polymer concentration regimes, rising salinity caused a reduction of the  $D_h$ . The increase of salinity (NaCl) from

0 to 500mg/L decrease the  $D_h$  from 770nm to 298 nm, at 25 °C, in dilute regime of HPAM concentration (10mg/L). The impact of salinity increase in  $D_h$  of HPAM concentrated dispersions is little significant, as well as the salinity increase from 500mg/L to 38000mg/L, in both concentration regime, (dilute and concentrated) at 25 °C.

### Temperature influence

The temperature increase from 25 °C to 50 °C promotes intensity increase in  $D_h$  of HPAM dilute dispersions. The  $D_h$  of 10 mg/L HPAM dispersion (dilute regime) in saline solution (500mg/L) increase from 298nm to 2500nm in this temperature range. This temperature difference was little significant in the  $D_h$  of HPAM concentrated dispersions. The increase temperature increase intensity the  $D_h$  of HPAM dilute dispersion and prevalence over the  $D_h$  decrease due the increase salinity.

### Flow in Porous Media

#### Artificial metallic porous media

Silva & Lucas [10] tested the same HPAM in artificial metallic porous media. The porous medium had permeability of 60,000 mD, pore diameter of 15 microns and they were 100% saturated with saline solution. The dispersion of HPAM of 1000mg/L (concentrated regime and  $D_h \sim 71$  nm) flowing in the porous medium resulted in a higher resistance factor (RF), but did not change the permeability (residual resistance factor (RRF)=1). While the 10mg/L HPAM dispersion (dilute regime and  $D_h$  2500nm) flowing in similar porous medium promoted a lower resistance factor, but a significant residual resistance factor. The permeability reduction attributed to polymeric retention in the porous medium is the sum of the contributions of adsorption, mechanical and hydrodynamics retentions [9]. In this metallic porous media, adsorption is not expected. The hydrodynamic retention is associated to the flow and is considered to be of little significance. However, the porous media were similar and the flow rate was the same. Then, the permeability reduction could be attributed primarily to mechanical retention. The greater retention obtained with the lower concentration polymer dispersion was due to its higher hydrodynamic volume.

#### Natural sandstone cores

High reduction of water permeability due to flow of HPAM diluted dispersions was also observed in natural porous media in bench tests. They were also 100% saturated with saline solution.

#### Injectivity tests in field applications

As reported in the literature, an injectivity test in the field, with this same HPAM showed higher RRF where there was higher water production before polymer dispersion application [7,8]. This behavior was detected by pressure monitoring and tracer application. In this case the reduction occurred in the water effective permeability, since the porous medium (reservoir) was naturally saturated with oil and water. The mechanism of this

result can also be credited to the expansion of the polymeric hydrodynamic volume due to dispersion dilution in diluted regime.

### Conclusion

The results obtained in the analyzed conditions show that the polymeric Dh expansion due to concentration reduction in the diluted regime and the temperature increase can promote significant mechanical retention, even in high salinity conditions. The mechanism presented favors the application of polymer flooding process in mature fields with high heterogeneity and salinity, of most economical form. The impact of concentration and temperature variation in the Dh can be expected for other flexible macromolecule systems and porous media.

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DOI: [10.19080/RAPSCI.2018.04.555637](https://doi.org/10.19080/RAPSCI.2018.04.555637)

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