



Overview on the Mechanism of Action of Two Well-known Antibacterial Drugs Based on Enzyme Inhibition



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Abstract

Most of the drugs achieve their therapeutic effects by interacting with specific target enzymes of the humans or infectants. The action of such drugs involves enzyme inhibition which can be reversible or irreversible. In this review, the discovery, structure, mechanisms of action of two important drugs based on enzyme inhibition are discussed. Among them, Penicillins are irreversible inhibitors whereas Sulphonamides are reversible inhibitors.

Keywords: Enzyme inhibitors; Penicillin; Sulphonamides; Drugs

Abbreviations: NAM: N-acetylmuramic acid; NAG: N-acetylglucosamine; SAR: Structure-Activity-Relationship

Introduction

During this pandemic, chemistry shines the brightest light on our path in the form of medicines. Drug is a chemical substance which has a physiological effect for treating diseases when ingested or administered into the body. From the times of Paul Ehrlich, father of modern chemotherapy, understanding the mechanism of action of drugs has been one of the key interests of scientific community. The journey of drug development starts with identifying lead compound, which may act on specific genes or proteins involved in a disease. The analysis of Structure-Activity-Relationship (SAR) enables the determination of the chemical group responsible for evoking a target biological effect in the organism. A drug achieves its therapeutic effects by interacting with specific target. These targets are the important biochemical components of the humans or infectants. Among these drug targets, enzymes are the very basic ones. Enzymes catalyze chemical reactions in the biological mediums. Binding of the substrates to its active site with the help of nonbonding interactions, is the main feature of an enzyme catalyzed reaction [1]. These interactions are balanced so that they can hold the substrates and allow products to leave the binding site, more over an enzyme stabilizes transition state

of the reaction by binding it most strongly [2]. Selective inhibition of these enzymes leads to the desired effects of a drug. Inhibition of an enzyme can be achieved in many different ways. Types of inhibitors which play a significant role in drug discovery processes are discussed below [3].

Reversible inhibitors

This type of inhibitors bind to the enzymes by non-covalent interaction, such as H-bonding, salt-bridge, van der Waals interaction and the enzyme inhibition can be reversed if the inhibitor is removed.

Drugs which are designed to be similar to the natural substrate, can fit to the active site more strongly than the substrate itself. It may not undergo any reaction when it is in the active site, but as long as it stays there it blocks access to the natural substrate and prevents the enzymatic reaction. This is known as Competitive inhibition. Uncompetitive inhibitor binds exclusively with enzyme-substrate complex. Non-competitive inhibitor binds at an allosteric site and distorts the active site. As long as the non-competitive inhibitor is bound, the enzyme remains catalytically inactive.

Irreversible inhibitors

Irreversible enzyme inhibitors can form a covalent bond to a key amino acid in the active site and permanently block the action of the target enzyme. These cannot be affected by substrate concentration. Sometimes, the target enzyme commits suicide by

activating the drug, so they are also known as suicidal substances.

Based on enzyme inhibition a vast number of drugs have been developed. Therapeutic uses of some drugs based on enzyme inhibition are listed in Table 1.

Table 1:

Enzyme Type		Drug Name	Target Enzyme	Disease
Reversible Inhibitor	Competitive	Statins- Atorvastatin, simvastatin	HMG CoA reductase	Decrease Plasma Cholesterol level – Antihyperlipidemic agents
		Allopurinol	Xanthine oxidase	Gout
		Sulphonamide	Transpeptidase	bactericidal
		Methotrexate	Dihydrofolate reductase	Cancer
		Dicoumarol	Vit.K-epoxide-reductase	Anti-coagulant
	Non-competitive	Phenelzine	Mono-amine oxidase (MNO)	Anti-depressants
	Uncompetitive	Camptothecin	Topoisomerase I	Cancer
Ciglitazone		Hepatic cytochrome P450 enzyme system	Inflammatory diseases	
Irreversible Inhibitor		Penicillin	Transpeptidase	Bactericidal
		Acyclovir	DNA-Polymerase	Antiviral
		Disulfiram	Alcohol dehydrogenase	Alcoholism

Among this huge variety of drugs, in this article, we are presenting a brief review on two representative drugs: Penicillin as irreversible inhibitor and Sulphonamide as reversible inhibitor. Our objective is to study their structure, function of the target enzyme and mechanism of drug action on that enzyme.

Penicillin

Background

In 1928 Scottish Scientist Alexander Fleming discovered Penicillin accidentally [4]. He observed that a blue-green mold of fungus had developed on a staphylococcus culture plate and the fungal colonies killed nearby bacterial colonies. He identified the fungus as a rare species of penicillium notetum [5]. But he could not separate the active compound and concluded that it is too unstable to isolate. In 1938, this problem was solved by Howard Florey and Ernst Chain. They developed the technology for large-

scale production of the active component produced by the mold. Bio-technological production of Penicillin revolutionized the medicinal field in the 20th century. Penicillin was available in the market as an antimicrobial drug saving the lives of many soldiers during Second World War. The Nobel Prize in Medicine in 1945 was awarded to Alexander Fleming along with Howard Florey and Ernst Chain for their contribution in creating the first mass-produced antibiotic.

In 1945, Dorothy Hodgkin established the structure of Penicillin using X-ray Crystallography (Figure 1). Penicillins are part of a broader class of antibiotics known as beta-lactam antibiotics. Its structure consists of a fused thiazolidine ring and a β lactum ring (cyclic amide). Penicillins are used to treat certain infections caused by bacteria such as pneumonia and other respiratory tract infections, scarlet fever, and ear, skin, gum, mouth, and throat infections.

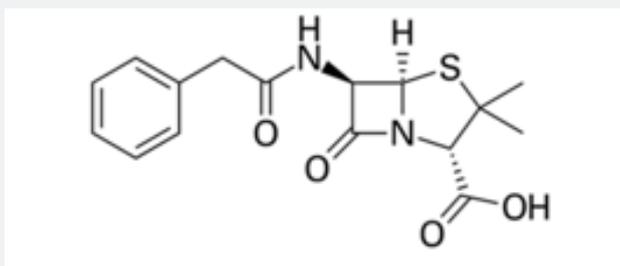


Figure 1: Structure of Penicillin G.

Function of target enzyme

Bacteria have cell wall in order to survive a range of environmental conditions. Without cell wall, water enters to the bacterial cell, lysis takes place. The wall is a peptidoglycan structure, made up of peptide and sugar units. The structure of the wall consists of a parallel series of sugar backbones containing two types of sugar, namely N-acetylmuramic acid (NAM) and N-acetylglucosamine (NAG). Peptide chains are bound to the NAM sugars and both D & L- amino acid present in this chain. In the final stage, the peptide chains are linked together by the displacement of d-alanine from one chain by glycine in another. This cross-linking reaction is catalysed by transpeptidase enzyme [6].

Transpeptidase has serine residue at the active site, which catalyzes the hydrolysis of peptide bonds (Fig. 2). Serine acts as a nucleophile to split the peptide bond between the two d-alanine units on a peptide chain. The terminal alanine departs the active site, leaving the peptide chain bound. The pentaglycyl moiety of another peptide chain now enters the active site and the terminal

glycine forms a peptide bond to the alanine group, displacing it from serine and linking the two chains together [7].

Mechanism of drug action

Penicillin works by inhibiting transpeptidase enzyme, which is involved in the last step of cell wall synthesis. Penicillin is the structural analogue of the d-Ala-d-Ala moiety. So, it binds to the enzyme. Once bound, penicillin is subjected to nucleophilic attack by serine (Figure 2). The enzyme can attack the β -lactam ring of penicillin and cleave it in the same way as it did with the peptide bond [8]. Penicillin is cyclic so the molecule is not split in two and nothing leaves the active site. Subsequent hydrolysis of the ester group linking the penicillin to the active site does not take place either, as the penicillin structure blocks access to the pentaglycine chain or water. In this way Penicillin binds with transpeptidase enzyme and inhibits it irreversibly. Since Penicillin is able to kill the bacterial cell, it is classed as bactericidal. Human cell remains unaffected by penicillin because human cells do not contain peptidoglycan.

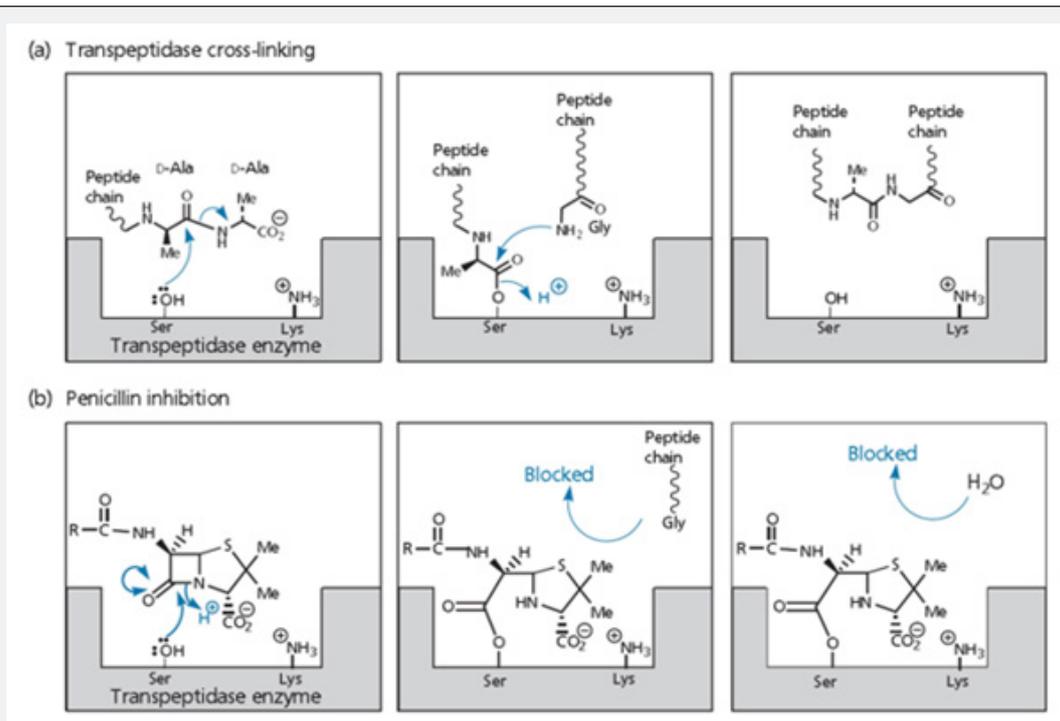


Figure 2: Mechanisms of transpeptidase cross-linking and penicillin inhibition.

Penicillin resistance

The overuse and misuse of Penicillin have led to a major problem of antibiotic resistance. Certain Bacteria gain penicillin resistance by the presence of β -lactamase enzymes which are the mutated form of transpeptidases (Figure 3). They can hydrolyse β -lactam ring of penicillin, thus making it inactive [9]. Also, some gram-negative bacteria produce excess amount of transpeptidase

enzyme and Penicillin is incapable of inactivating all the enzyme molecules present.

Successful derivatives & broad spectrum penicillins

Those penicillins having useful activity against both Gram-positive and Gram-negative bacteria are known as broad-spectrum antibiotics. Examples are Ampicillin, Amoxicillin, Carbenicillin,

Ureidopenicillins. These are the first line of defence of any bacterial infection [10]. Many successful derivatives of penicillin

are developed by changing the 'R'-group (shown in Figure 3). Some of them are listed in Table 2.

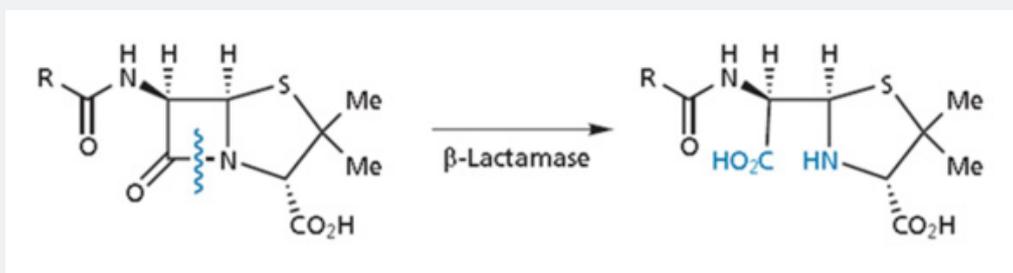
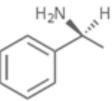
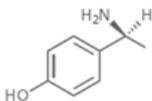
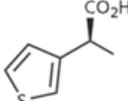
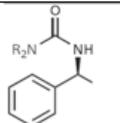
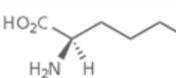
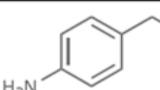


Figure 3: β -lactamase deactivation of penicillin.

Table 2:

R-group	Drug Name
	Ampicillin
	Amoxicillin
	Ticarcillin
	Ureidopenicillins.
	Penicillin N
	Penicillin T

Sulphonamide

Background

In 1935, German bacteriologist and pathologist Gerhard Domagk discovered that a red dye Prontosil had in vivo antibacterial properties against fatal streptococci infections in humans. But strangely no antibacterial effect was observed in vitro [11]. This remained a mystery until French scientist Ernest Fourneau discovered that prontosil was metabolized by bacteria present in the small intestine of the test animal to give a product called Sulphanilamide (Figure 4). It was this compound which was the true antibacterial agent [12]. Domagk got Nobel Prize in 1939 for development of sulfonamides such as Prontosil. Sulphonamides are a group of synthetic medicines that contain

the sulphonamide chemical group (Figure 5). They may also be called sulpha drugs. They are used to treat urinary tract infections, infections of mucous membranes, gut infections, etc.

Function of target enzyme

Bacteria require dihydropteroate for the synthesis of tetrahydrofolate (Coenzyme-F), which is necessary for DNA synthesis of bacteria (Figure 6). Para-aminobenzoic acid (PABA) first binds with the enzyme dihydropteroate synthetase and then enzyme-substrate complex reacts with 6-hydroxymethylpterin diphosphate to produce dihydropteroate, which reacts with L-glutamic acid to form dihydrofolate. Dihydrofolate again binds with dihydrofolate reductase enzyme to produce tetrahydrofolate (Coenzyme F). Now coenzyme-F is used for DNA synthesis.

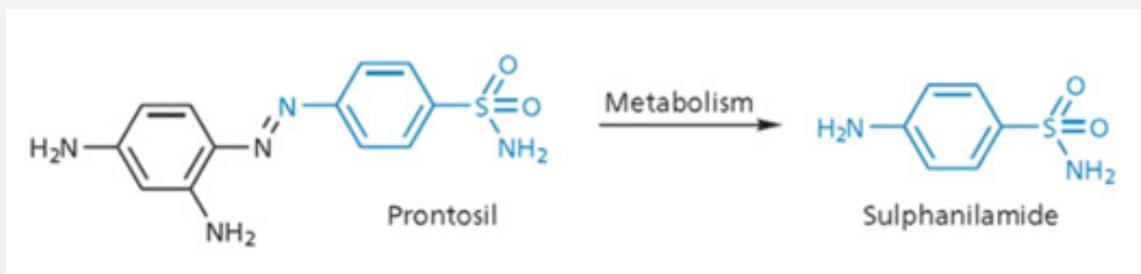


Figure 4: Activation of Prontosil to form Sulphanilamide.

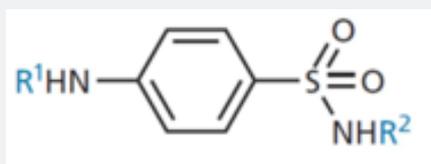


Figure 5: General Structure of Sulphonamides.

Mechanism of drug action

Sulphonamides are the example of competitive enzyme inhibitors and structural analogues of para-aminobenzoic acid (PABA) [13]. Sulphonamide binds with dihydropteroate synthase enzyme instead of PABA and inhibits it reversibly (Figure 6). So, the resulting enzyme-substrate complex does not undergo further enzymatic reaction. In this way, Sulphonamide blocks the biosynthesis of tetrahydrofolate which is required for DNA synthesis of bacteria. Now, bacteria cannot grow, and our human immunity system gets sufficient time to kill the bacteria. Sulphonamide does not actively kill the bacteria, only prevent the cell growing and multiplying, hence it is classed as bacteriostatic.

These drugs are further classified as antimetabolites since they prevent the cell metabolism of bacterial cell.

Selectivity of sulphonamide drug

Tetrahydrofolate is also necessary for human cell. But sulphonamide is not toxic for human body. This is because human cell can synthesis tetrahydrofolate from folic acid - a totally different pathway. This folic acid is obtained from the diet as a vitamin and then transport to human cell by transport protein. Bacterial cell is unable to follow the same mechanism of human cell due to lack of transport protein. The success of sulphonamides is based on the utilization of metabolic differences between mammalian and bacterial cells.

Sequential blocking

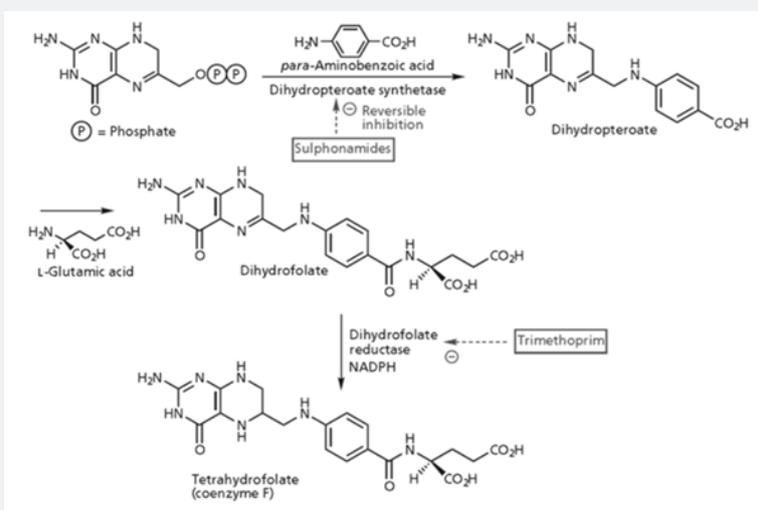


Figure 6: Production of Coenzyme F and its inhibition by sulphonamides.

As sulphonamides are reversible inhibitors, their efficiency depends on the substrate concentration but administration of large amounts these drugs produced toxic effects on the patients. This drawback is solved by sequential blocking of two enzymes required in a biosynthetic pathway [14]. For example trimethoprim which blocks Dihydrofolate reductase, is used with sulphonamides to produce antibiotic effect of higher strength (Figure 6).

Successful derivatives

Sulfanilamide itself has been superseded in medicines by its derivatives. Structure-activity relationships reveal that the fruitful derivatives have heterocyclic ring as substituent in place of one hydrogen of sulphonamide group (-SO₂NHR₂, Figure 5). Some of them are listed in Table 3.

Table 3:

Structure	Name of drug
	<u>Sulphadoxine</u>
	<u>Sulphamethoxazole</u>
	Sulfadiazine
	<u>Sulphamethazine</u>
	<u>Sulphamethoxypyridazine</u>

Conclusion

Enzymes remain prime focus for current drug design because altering enzyme activity has immediate and defined effects. In this review, we have discussed discovery, structure, mechanisms of action of two important drugs based on enzyme inhibition. Penicillin which is irreversible inhibitor of transpeptidase enzyme can kill the bacterial cell. On the other hand Sulphonamides which are reversible inhibitors prevent the cell metabolism of bacterial cell. Profound understanding of the target enzyme function and course of an enzyme-catalysed reaction can help to conceptualize different types of inhibitors. Currently, 47% of total drugs in the world are based on enzyme inhibition. With the help of computer modelling and spectroscopic tools the drug design and development will reach new heights at near future.

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