

Structural and Electrical Properties of Multivalent Pr Doped Ceria: A Short Review

Sk. Anirban*

Department of Physics, Govt. General Degree College, Salboni, India

Submission: March 12, 2017; Published: March 31, 2017

*Corresponding author: : Sk. Anirban, Department of Physics, Govt. General Degree College, Salboni, India; Email: sk.anirban@gmail.com

Abstract

Multivalent Pr doped ceria is a potential candidate for SOFCs due to its mixed electronic and ionic conductivity. The valence state of Pr significantly affects the lattice parameter of Pr doped ceria. The lattice parameter mainly depends on the dopant concentration, sintering and switching of valence state. The concentration oxygen vacancies are responsible for ionic conductivity and the oxidation of Pr³⁺ to Pr⁴⁺ is the origin of electronic conductivity. The electronic conductivity increases with the doping concentration and the oxygen partial pressure.

Keywords: Pr doped ceria; Mixed conductivity; Oxygen partial pressure

Introduction

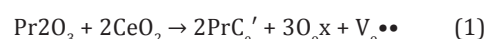
Cubic fluorite structured ceria doped with lower valent cations forms exceedingly oxygen vacancies in the doped ceria lattice. Therefore rare earth doped ceria exhibits high oxygen conductivity at elevated temperatures, which makes it potential candidate for solid electrolyte in intermediate temperature solid oxide fuel cells (IT-SOFCs) [1]. Among the different rare earth elements, multivalent Pr draws more attraction due to its ease of Pr³⁺/Pr⁴⁺ switching, similar structure to that of ceria and ionic radius closed to Ce⁴⁺ ions, which make Pr doped ceria as oxygen storage materials and oxygen ion conducting solid electrolytes [2,3]. Again, Pr doped ceria exhibits pink-orange to red-brown colored nano-particles depending on the synthesis process, sintering temperature, dopant concentration, etc. [4]. Takasu et al. [5] reported that multivalent Pr doped ceria can be used as an oxide electrode due to its high electronic and ionic conductivity. Again Ahnet al. [6] reported that Pr³⁺ have been included as preponderance ions in Pr-doped ceria which indicates higher ionic conductivity than electronic conductivity in Pr doped ceria. However, electronic conductivity is due to small polar hopping of electrons between Pr³⁺ and Pr⁴⁺ ions and the ionic conductivity is due to the formation of oxygen vacancies [7]. In this mini review, structure and mixed conductivity of Pr doped ceria have been discussed.

Discussion

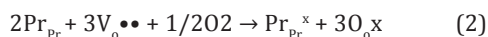
The single phase cubic fluorite structure with space group Fm3m of Pr doped ceria has been confirmed from the XRD

patterns by the several researchers [4,8-10]. de Larramendi IR, et al. [10] reported that the lattice parameters of Pr doped ceria as 5.4117(4), 5.4097(9) and 5.4234 (1) Å for the samples prepared by ceramic, sol-gel and modified polyol routes, respectively. The lattice parameters was found to increase with Pr concentration [4,8] which is due to the higher ionic radius of Pr³⁺ (1.126 Å) than Ce⁴⁺ (0.97 Å). Since ionic radius of Pr⁴⁺ (0.94 Å) is slightly lower than Ce⁴⁺ thus lattice expansion due to Pr doping indicates Pr³⁺ is the major dopant cautions. According to Chiba R, et al. [9] lattice parameter of Pr doped ceria decreases with dopant concentration at lower temperature but increases at higher temperature. Y. Takasu et al [11] also reported decreases in lattice parameter with Pr concentration. The contradiction in results may be due to the different synthesis route which is responsible for different Pr³⁺/Pr⁴⁺ ratio in the formed ceria based solid solutions. However, the rate of increase of lattice parameter with dopant concentration depends on Pr³⁺/Pr⁴⁺ switching with the sintering temperature [8].

The ionic conductivity in Pr doped ceria depends on the formation of free oxygen vacancies due to the substitution of Ce⁴⁺ ions by Pr³⁺ which is represented by the Kröger-Vink notation as:



Since Pr is multivalent the concentration of free oxygen vacancies may decrease due to the oxidation of Pr³⁺ to Pr⁴⁺ as represented by



Therefore with the Pr concentration the ratio of ionic to electronic conductivity decreases in Pr doped ceria. Shunk, et al. [12] reported that the electronic conductivity exceeds the ionic conductivity at doping concentration >25%. It is also reported that at very low dopant concentration (<2%), Pr doped ceria exhibits p-type electronic conduction at 0.2atm or a high oxygen partial pressure but at high doping concentration it exhibits n-type electronic conduction [9]. Anirban SK, et al. [8] has been investigated the sintering effect on Pr doped ceria and was found fall in ionic conductivity at higher sintering temperature due to the oxidation of Pr³⁺ to Pr⁴⁺ and formation of defect association which is reduces the free oxygen vacancies. The oxygen vacancy and electron concentration in Pr doped ceria is very sensitive to temperatures at which conductivity changes abruptly and this temperature decreases with doping concentration [9]. Using kinetic lattice Monte Carlo model for oxygen vacancy diffusion in Pr doped ceria Dholabhai PP, et al. [13] observed maximum in ionic conductivity at 15-30% dopant concentration. Again the conductivity of multivalent Pr doped ceria strongly depends on the oxygen partial pressure, unlike than other trivalent doped ceria in which the ionic conductivity is independent of oxygen partial pressure. The oxidation of Pr³⁺ to Pr⁴⁺ increases with oxygen partial pressure and at sufficiently high oxygen partial pressure most of the Pr takes on the +4 state and ionic conductivity exhibits a power law dependence on oxygen partial pressure with a -1/6 slope [14]. The electronic conductivity of Pr doped ceria predominant at higher oxygen partial pressure. Since Pr is mixed valent thus the small polar on hopping of electrons on Pr³⁺ sites can only hope to a neighboring empty Pr site or in the Pr⁴⁺ state. Thus the electronic conductivity due to small polar on is proportional to the product Pr³⁺ and Pr⁴⁺ concentration and shows a maximum conductivity as a function of oxygen partial pressure where the concentrations of Pr³⁺ and Pr⁴⁺ are equal.

Conclusion

The valence state of Pr in Pr doped ceria depends on the synthesis process, sintering temperature and oxygen partial pressure. Depending on the valence state of Pr the lattice parameter of Pr doped ceria may increase or decrease. The oxidation of Pr suppresses the oxygen vacancies and increases the electronic conductivity. The values of mixed electronic-ionic conductivity depend on the dopant concentration and oxygen

partial pressure. The maximum conductivity has been obtained at the equal concentration of each valence state of Pr in Pr doped ceria.

References

1. Inaba H, Tagawa H (1996) Ceria-based solid electrolytes. *Solid State Ionics* 83(1-2): 1-16.
2. DL Maricle, TE Swarr, S Karavolis (1992) Enhanced Ceria-A Low-Temperature SOFC Electrolyte. *Solid State Ionics* 52(1-3): 173-182.
3. P Knauth, HL Tuller (1999) Non-stoichiometry and Relaxation Kinetics of Nanocrystalline Mixed Praseodymium-Cerium Oxide Pr_{0.7}Ce_{0.3}O_{2-x}. *J Eur Ceram Soc* 19(6-7): 831-836.
4. Aruna ST, Ghosh S, Patil KC (2001) Combustion synthesis and properties of Ce_{1-x}Pr_xO_{2-δ} red ceramic pigments. *Int J Inorg Mater* 3(4-5): 387-392.
5. Takasu, Sugino T, Matsuda Y (1984) Electrical conductivity of praseodymia doped ceria. *J Appl Electrochem* 14(1): 79-81.
6. Ahn K, Yoo DS, Prasad DH, Lee HW, Chung YC (2012) Role of Multivalent Pr in the Formation and Migration of Oxygen Vacancy in Pr-Doped Ceria: Experimental and First-Principles Investigations. *Chem Mate*. 24(1): 4261-4267.
7. Bishop SR, Chen D, Kuru Y, Kim JJ, Stefanik TS, et al. (2011) Measurement and Modeling of Electrical, Mechanical, and Chemical Properties of a Model Mixed Ionic Electronic Conductor: Pr Doped Ceria. *ECS Trans* 33(40): 51-57.
8. Anirban SK, Dutta A (2017) Structural interpretation of optical properties and ion transport mechanism in mixed valent Pr containing nanoceria. *Mat Res Bull* 86: 119-130.
9. Chiba R, Taguchi H, Komatsu T, Orui H, Nozawa N, et al. (2011) High temperature properties of Ce_{1-x}Pr_xO_{2-δ} as an active layer material for SOFC cathodes. *Solid State Ionics* 197(1): 42-48.
10. de Larramendi IR, Ortiz-Vitoriano N, Acebedo B, de Aberasturi DJ, de Muro IG, et al. (2011) Pr-doped ceria nano particles as intermediate temperature ionic conductors. *Int J Hydro Energy* 36: 10981-10990.
11. Takasu Y, Sugino T, Matsuda Y (1984) Electrical conductivity of praseodymia doped ceria, *J Appl Electrochem* 14(1): 79-81.
12. Shuk P, Greenblatt M (1999) Hydro thermal synthesis and properties of mixed conductors based on Ce_{1-x}Pr_xO_{2-δ} solid solutions. *Solid State Ionics* 116(3-4): 217-223.
13. Dholabhai PP, Anwar S, Adams JB, Crozier P, Sharma R (2011) Kinetic lattice Monte Carlo model for oxygen vacancy diffusion in praseodymium doped ceria: Applications to materials design. *J Solid State Chem* 184(4): 811-817.
14. Tuller HL, Bishop SR, Chen D, Kuru Y, Kim YY, et al. (2012) Praseodymium doped ceria: Model mixed ionic electronic conductor with coupled electrical, optical, mechanical and chemical properties. *Solid State Ionics* 225: 194-197.



This work is licensed under Creative Commons Attribution 4.0 License
DOI: 10.19080/JOJMS.2017.01.555551

Your next submission with Juniper Publishers will reach you the below assets

- Quality Editorial service
- Swift Peer Review
- Reprints availability
- E-prints Service
- Manuscript Podcast for convenient understanding
- Global attainment for your research
- Manuscript accessibility in different formats (**Pdf, E-pub, Full Text, Audio**)
- Unceasing customer service

Track the below URL for one-step submission
<https://juniperpublishers.com/online-submission.php>