Self-Micro-Emulsifying Drug Delivery System to Enhance the Solubility of the Hydrophobic Drugs

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Abstract

Self-micro-emulsifying drug delivery system is one of the approach for enhancing the solubility of the hydrophobic drug. The drugs which are insoluble in water can be formulated in this method by solubilising it in the lipid vehicle to absorb through the membranes. The lipid and surfactants are used to increase the solubility of the drug and improve absorption. This improves the dissolution rate of the drug by increasing its solubility. Many of the hurdles and solutions are described in this article. This technique gained attention as it also enhances the bioavailability of the drug. This article has complete review about SMEDDS for present work and for future perspective also.

Keywords: Hydrophobic; Lipid; Surfactants; Dissolution; Bioavailability

Introduction

Oral route is the most preferred, affordable and convenient route to administer a drug. The pathways through which the drug distributed can be through gastro-intestinal tract or lymphatic system. They are preferable as they are easy to administer, low cost, and low patient compliance. Oral route mainly focuses on some of the main absorption sites of the body which is mouth, stomach, intestine, rectum etc. some of the dosage forms are solid oral dosage forms like tablets, capsule, pellets etc. and on the other hand liquid oral dosage forms like emulsion, suspension, gels etc. As of the poor patient compliance and difficult to administer the drug to paediatric and geriatric patients, fast dissolving techniques are introduced to administer the drug for immediate action and for better patient compliance. The dosage form includes the tablets and films. The dosage forms that are available are oral disintegrating tablet, mouth dissolving tablet, mouth melting tablet, oral disintegrating film, and mouth melting wafer, mouth dissolving film and sublingual tablet or film.

The fast dissolving technology is one of the oral route dosage form technique eases the transport of drugs to act immediately. They permeate through interstitial spaces by the mucosa and directly go into the systemic circulation for the immediate action. They disintegrate rapidly and dissolve in saliva and absorb through the mucosal surface.

Advantages of fast dissolving dosage forms:

1. Self-medication
2. No first pass metabolism
3. No use of water
4. Improved patient compliance
5. Ease of manufacturing
6. Lack of pain
7. Immediate action
8. Easy for pediatric and geriatric patients
9. Ease of administration (mentally ill, disabled, uncooperative)
10. Leaves no residue on mouth after administration
11. Faster absorption
12. Cost effectiveness

The drugs which are hydrophilic (soluble in aqueous solvents) are considered for the fast dissolving techniques but for those which are lipophilic cannot be formulated as they are insoluble in aqueous solvent and soluble in non-aqueous solvents. The lipid formulations can enhance the solubility of the hydrophobic drug are liposomes, solid lipid nano-particles, self-dispersing tablets, solid dispersions etc. the drug formulated should be permeable across the gastric mucosa and should be least sparingly soluble in water (BCS class II and IV).
SMEDDS

The solubility of hydrophobic drugs can be enhanced by various methods and one of them are self-micro-emulsifying drug delivery system (SMEDDS) or self-emulsifying drug delivery system (SEDDS) as they also enhance the permeability and bioavailability of the hydrophobic drugs without going into first pass metabolism [1,2]. Self-micro-emulsifying drug delivery systems (SMEDDS) are the isotropic mixture and lipid based formulations given by oral route. The hydrophobic drug which is soluble in the oil can be formulated in this technique. The mixture consists of oil which has to be in the range of less than 100nm droplet size. The bioavailability of the hydrophobic drug will be enhanced in situ solubilisation of the drug absorbed by the lymphatic pathway. The SMEDDS will get in contact with the GIT lumen and forms w/o micro emulsion with the GI fluid. This method is advantageous for the poorly water-soluble drugs with the lipid as vehicle and surfactants to form w/o micro emulsion within the GIT lumen. For example-Fexofenadine [3], domperidone [4], danazol [5] etc.,

The first purpose to use the lipid is to solubilise the drug and the second one is to stimulate the biliary and pancreatic secretion which helps for the digestion of the lipids. The enzyme presents in the secretions act at the lipid/water interface and forms the mixed micelle which helps in the drug solubilisation [6]. Highly lipophilic drug (log P>5) show high solubility in the triglycerides (>50mg/L) which is good for lymphatic pathways. In the cellular level, the lipids open the tight junctions in the intestine which increases the permeability only for the class IV drugs (low solubility and low permeability) [7-9].

Advantages

1) Increase bioavailability by decrease in dose.
2) Decrease in food effects.
3) Prolong release of medicament.
4) For both liquid and solid dosage forms.
5) Overcome of the irritation caused by the contact between drug and the wall of GIT.
6) Simple mixing.
7) Low energy consumption.
8) Easy manufacturing process.
9) Less time required.
10) Hydrophobic drugs can be absorbed by stable plasma-time profile.
11) Drugs which can be degraded by GIT can be used.

Composition

1. Lipids [10,11]
2. Surfactants [12]
3. Co-surfactants [12]

Lipids

It solubilizes the drug and access to the lymphatic circulation for the absorption. They are used in the concentration of less than 20% w/w of the formulation. The lipids that are available are digestive lipids such as triglyceride, diglyceride, fatty acids, phospholipids and cholesterol improves the bioavailability of the drug [14]. The non-digestive ones are not used as they decrease the bio-availability and give impairment of the absorption. Edible oils are not useful as they give no solubilization of the drug and require large molecular volume. For self-micro-emulsifying drug delivery system, triglyceride with long and medium chain length (LCT and MCT) [15].

The medium chain triglyceride (MCT) [16]:
1) Digestible
2) Greater fluidity
3) Increase solubility
4) Good emulsifying property
5) Minimizes oxidation
6) Increase drug absorption
7) Positive effect on bioavailability
8) No access to lymphatic system

For example – re-esterification of fractionated coconut oil fatty acid with glycerin, palm seed oil, etc.

Long chain triglyceride (LCT)
1) Vegetable source
2) Enhance the lymphatic pathways
3) Safe for the use
4) Increase bioavailability

For example-cottonseed oil, soya bean oil, safflower oil, olive oil, peanut oil, sesame oil, wheat germ oil, rice bran oil, corn oil and castor oil.

Surfactants

They are amphiphilic in nature and HLB more than 12 are considered for the method which are water soluble. The concentration range taken should be between 30-60% w/w of the formulation.

1) Give self-emulsification property
2) Solubilizes the hydrophobic drug
3) Dissolution rate can be improved
4) In vivo Inhibitory effect on drug precipitation
5) Improves permeability by opening the tight junctions
6) Stabilizes the micro emulsion system

Selection of surfactant are based on HLB value, if the HLB value are high then it facilitates o/w microemulsion and if HLB>12 increases the solvent capacity of the formulations in which the droplet size will be <100nm.

Natural surfactant such as lecithin are less toxic but have limited self-emulsification efficiency. Anionic surfactants such as potassium laurate and sodium lauryl sulphate are used and in cationic surfactant, quaternary ammonium is used.

The best matches for the formulation are non ionic surfactants which are,

1) High HLB value
2) Less toxic than ionic
3) Greater emulsion stability

For example, oleates, polysorbates (tween 40, 60 and 80), polyoxyls which has HLB value 2 to 18 [17,18]. Disadvantages include GI irritation, low emulsification efficiency, dehydration effect on SGC and HGC. The combination of non-ionic and ionic surfactant gives increase in microemulsion, but it gives possible synergism.

Co-solvents and co-surfactants

They should be hydrophilic in nature.

Purpose

1) Access the entry of water into the formulation
2) Assist the dispersion process
3) Imparts the flexibility to the interface

The range of co-solvents and co-surfactants should be in 20-50% w/w of the formulation and always 30% w/w more than the concentration of the surfactants. The HLB value of the surfactants should be 10-14. Alcoholic surfactants are not used in the formulation as they cause precipitation of the drug. Short chain alcohols are the best to formulate in SMEDDS which are ethanol, polypropylene glycol, polyethylene glycol.

Drug candidate

1) Solubility: The drug should be insoluble in water. The log P should be greater than 4 (lipophilic), high solubility in LCT for lymphatic absorption (>50mg/ml).

BCS class II and IV [19]

2) Low melting point: The m.p should be low for the better absorbability of the drug [24].

3) Should be chemically and physically stable in SMEDDS.

Formulation

Screening of excipients

1) Solubility studies:

Method: Shake flask method in which the drug is solubilized with the excipients (excess amount) 0.5gm each of the solvent. It is shaken for the 48hrs a water bath shaker at room temperature. The solution is then centrifuged for 15 minutes. Then the supernant is filtered (0.45um). Then the drug content will be determined by HPLC [25-31].

2) Screening of surfactants and co-surfactants for their self-emulsification ability: Equal proportions of oil and surfactants (0.3gm) are homogenized for 2 min and then warmed for 30secs (40-45 °C). Then the 50mg of prepared isotropic mixture is added to distilled water and allowed to stand for 2hrs. The prepared micro emulsion are filtered and tested for clarity, turbidity and % transmittance (wavelength-638nm). If the resultant % transmittance is high then the emulsification efficiency will be high, low flask inversions, clear emulsion if formed then those surfactants and co-surfactants are selected.

Construction of pseudoternary phase diagram

Co surfactant + surfactant = Smix

It helps in deciding the ratio of oil, surfactant and co-surfactant to be formulated in the SMEDDS by pseudo ternary diagram. The ratio of oil, surfactant and co surfactant are to be added in vials with 5% w/w of water. They are centrifuged for 2-3min for phase separation. They are incubated for 48hrs (25 °C). The vials are shaken and the ratio of formulation which shows clear and isotropic mixture are selected [32,33].

Preparation

The selected hydrophobic drug, oil, surfactant, and co-surfactant are vortexed for 5-10min in a magnetic stirrer. The prepared emulsion is to be placed in oven at 50 °C for 1hr. Then it is checked for turbidity. If required vortex for 48 hours until clear solution obtained or heat it [34-37].

Characterization

a) Visual evaluation: The prepared SMEDDS are dissolved in water, if the emulsion appears opaque or milky white then it is macro emulsion and if it appears clear, isotopic and transparent then it is micro emulsion [38].

b) Droplet size analysis: SMEDDS are dissolved in water and tested for scanning electron microscope (SEM) which ranges 10 to 5000nm. The size should be narrow and less than 100nm.
c) Zeta Potential Measurement: The SMEDDS are diluted and checked for stability under Zeta potential analyzer. Higher the zeta potential there will be good stability. If the zeta potential is negative, then there is presence of fatty acid if they are positive then there is presence of cationic lipids [39,40].

d) In-vitro dissolution time OR dispersibility test: Apparatus: USP type II paddle are used with 50rpm speed at 37 °C. The medium used 200ml of 0.1N HCL.

e) Cloud point determination: The prepared formulation (0.5ml) are added to the distilled water (50ml) are to be placed on water bath. The temperature is to be raised at the rate of 0.5 °C /min then the emulsion is cooled till they become cloudy and they are checked spectrophotometrically.

f) Viscosity measurement: Rheometers-Brookfield cone and plate rheometer with cone spindle/rotating spindle Brookfield viscometer are used.

g) Dilution studies: The smedds are diluted to 100 times in double distilled water, simulated gastric fluid (SGF) and simulated intestinal fluid (SIF). If there is increase in the dilution or the appearance is clear, then there is no drug precipitation in the formulation. The effect of pH also can be evaluated by the buffer pH 1.2, 6.8 and distilled water and should be observed for transparency.

h) Refractive index: The property which can evaluate the isotropic nature of the emulsion. Constant refractive index shows the thermodynamic stability of the formulation. They are measured with the refractometers and compared with water. It depends on the nature and amount of co-surfactant and the globule size of the formulation. Lower the refractive index, higher the co surfactant concentration, and lower the rigidity of micro emulsion.

i) % Transmittance: The formulation is added in water and checked spectrophotometrically. If it is near to 100% then the emulsion is clear and transparent.

j) Transmission electron microscopy (TEM): Investigates the structures and morphology of micro emulsion by dilution of SMEDDS.

k) Differential Scanning Colorimetry and NMR techniques: DSC gives information about water which is in free state and bound state. In NMR, Fourier transformed pulsed gradient spin echo method (PGSE) PGSE-NMR

l) Thermodynamic Stability Studies: This study is for stability and to evaluate the consequence of temperature change on formulations.

m) The formulation of SMEDDS are dissolved in aqueous phase and centrifuged for 15 minutes at 1500 RPM or 30 minutes at 3500RPM.

n) Freeze thawing method in which the formulation is kept at -4 °C for freezing (24 hours) and at 40 °C thawing (24 hours) after that centrifuged for 3000RPM (5Min).

o) Stability Assessment: According to ICH guidelines, sample of SEMDDS are dissolved in distill water at 2-5 °C in cooler and at room temperature.

p) % Drug Content: The samples of SMEDDS in 100ml volumetric flask are added with an extracting solvent. Then it is shaking for 1 hour [Mechanical Shaker] and kept aside for 24 hours. The sample is then filtered, and absorbances are taken in UV.

q) Phase separation study: 1ml sample of SMEDDS are taken with 5ml 0.1N HCl in a glass tube. Then the buffer pH 6.8 and distill water are added. They are altered for 3-4 times between 2 hours [40-43].

Conflict of Interest
Authors Declared None

References


