



# Ensuring Adequate Iron Status in Vegetarians and Vegans



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## Abstract

Many people, including some physicians, are concerned about iron deficiency anemia in patients consuming a plant-based diet. However, studies show that the risk of anemia in vegetarian and vegan patients is no greater than in omnivorous patients.

Plant foods contain only non-heme iron, whereas meat contains both the heme and non-heme iron. There used to be a concern that non-heme iron would be poorly absorbed resulting in iron deficiency anemia. However, non-heme is well absorbed in most vegetarian patients and vegan because other plant foods, containing substances such as vitamin C and citric acid, can greatly enhance its absorption. Furthermore, non-heme iron absorption increases whenever iron stores are low.

Adequate iron levels can easily be maintained in the vegan patient with a little planning. Consuming foods high in iron along with foods that enhance non-heme iron absorption, will prevent iron deficiency anemia in vegan and vegetarian patients. Because both groups of foods are widely available, this should not be difficult to accomplish.

Patients that are already anemic can be treated by increasing their consumption of iron rich and iron enhancing foods. Supplements are sometimes required. In these cases, iron supplements can be prescribed in the same manner as with omnivorous patients.

**Keywords:** Anemia; Deficiency; Heme iron; Hcpidin; Iron; Non-heme iron; Plant-based diet; Supplements; Vegetarian; Vegan; Plant foods; Enterocytes; Macrophages; Synaptogenesis; Myelination; Oxidative phosphorylation; Cytochromes, Sulfuric Proteins

## Introduction

### Iron sources

There are two types of iron in food: heme and non-heme iron. In animal products, 40% of the total iron content is heme iron and 60% non-heme iron [1]. Plant foods contain only non-heme iron.

Many breakfast cereals and some breads are also fortified with iron [2].

Plant foods that naturally contain iron, are wholegrain cereals and breads; dried beans and legumes; dark green leafy vegetables; dried fruits; and nuts and seeds [3] (Table 1).

**Table 1:** Sources of Iron in Plant Foods.

Food	Serving Size	Iron (mg)
<b>Vegetables and Fruit:</b>		
Spinach, cooked	125mL (1/2 cup)	2.0-3.4
Tomato puree	125mL (1/2 cup)	2.4
Edamame, baby soybeans, cooked	125mL (1/2 cup)	2.2
Asparagus, raw	6 spears	2.1
Hearts of palm, canned	125mL (1/2 cup)	2
Potato with skin, cooked	1 medium	1.3-1.9
Snow peas, cooked	125mL (1/2 cup)	1.7
Turnip or beet greens, cooked	125mL (1/2 cup)	1.5-1.7
Prune juice	125mL (1/2 cup)	1.6

Apricots, dried	60mL (1/4 cup)	1.6
Beets, canned	125mL (1/2 cup)	1.6
Kale, cooked	125mL (1/2 cup)	1.3
Green peas, cooked	125mL (1/2 cup)	1.3
Tomato sauce	125mL (1/2 cup)	1.3
<b>Grain Products:</b>		
Oatmeal, instant cooked	175mL (3/4 cup)	4.5-6.6
Cream of wheat, all types, cooked	175mL (3/4 cup)	5.7-5.8
Cereal, dry all types	30g (check product label for serving size)	4.0-4.3
Granola bar, oats, fruit and nut	32g (1 bar)	1.2-2.7
<b>Legumes, nuts and seeds:</b>		
Tofu, cooked	150g (3/4 cup)	2.4-8.0
Soybeans, mature, cooked	175mL (3/4 cup)	6.5
Lentils, cooked	175mL (3/4 cup)	4.1-4.9
Beans (white, kidney, navy, pinto, black, adzuki), cooked	175mL (3/4 cup)	2.6-4.9
Pumpkin or squash seeds, roasted	60mL (1/4 cup)	1.4-4.7
Peas (chickpeas, black-eyed, split), cooked	175mL (3/4 cup)	1.9-3.5
Tempeh/fermented soy product, cooked	150g (3/4 cup)	3.2
Meatless sausage, chicken, meatballs, fish sticks, cooked	75g (2.5 oz)	1.5-2.8
Baked beans, canned	175mL (3/4 cup)	2.2
Nuts (cashews, almonds, hazelnuts, macadamia, pistachio), without shell, raw	60mL (1/4 cup)	1.3-2.2
Sesame seeds, roasted	15mL (1 Tbsp)	1.4
Meatless luncheon slices	75g (2.5oz)	1.4
Hummus	60mL (1/4 cup)	1.5
Almond butter	30mL (2 Tbsp)	1.1

**Source:** Canadian Nutrient File 2015.

### Physiology and pathology

Iron is a transition metal and has multiple functions in more than 180 biochemical reactions in the human body, including electron transport in redox reactions (cytochromes, sulfuric proteins), redox catalytic functions (cytochrome p450, catalase, peroxidase) and reversible storage and transport of O<sub>2</sub> (hemoglobin, myoglobin). It also plays an important role in the production of neurotransmitters, and is essential in synaptogenesis and myelination. Oxidative phosphorylation is the most critical biochemical pathway in which iron is involved [4-6].

Iron deficiency anemia usually develops slowly. As iron levels decline in the stores (iron deficiency) and in the circulation (iron restricted erythropoiesis), becoming insufficient for the full hemoglobinization of mature erythroblasts (iron deficiency anemia), the liver peptide hormone hepcidin is transcriptionally suppressed [7]. Indeed, serum hepcidin levels are significantly lower in young women with a negative iron balance compared with males and postmenopausal women, and are even undetectable in serum of individuals with iron deficiency anemia [8-10]. The decrease of hepcidin enhances iron release into plasma through

ferroportin from both enterocytes and macrophages in the attempt to maintain normal transferrin [7].

### Absorption

The amount of non-heme iron absorbed is primarily determined by the body's need for iron - people with the lowest iron stores will absorb more and excrete less [2,11]. Humans can adapt successfully to a wide range of iron requirements and intakes [12]. If iron intake is low, vegetarians adapt by excreting less fecal ferritin [11]. In pregnant women who need the most iron, absorption can increase by 60% relative to normal [12,13]. Non-heme iron is nearly as well absorbed as heme iron in people with very low iron stores [12]. There is apparently no advantage in storing more than a minimal amount of iron [14].

Absorption is increased as much as three to six-fold with the addition of 50 mg of vitamin C per meal [3]. Vitamin C facilitates the conversion of Fe<sup>3+</sup> (ferric) to Fe<sup>2+</sup> (ferrous) iron, the form in which iron is best absorbed. This process is carefully regulated by the gut. Vegetarians typically have high intakes of vitamin C from a wide variety of fruits and vegetables.

Other common organic acids such as citric and malic [15], as well as vitamin A and  $\beta$ -carotene, also enhance non-heme iron absorption [15,16].

This Vitamin C table shows that even modest servings of fruits and vegetables supply adequate vitamin C to enhance iron absorption (Table 2).

**Table 2:** Selected Food Sources of Vitamin C.

Food	Milligrams (mg) per serving	Percent (%) DV*
Red pepper, sweet, raw, ½ cup	95	158
Orange juice, ¾ cup	93	155
Orange, 1 medium	70	117
Grapefruit juice, ¾ cup	70	117
Kiwifruit, 1 medium	64	107
Green pepper, sweet, raw, ½ cup	60	100
Broccoli, cooked, ½ cup	51	85
Strawberries, fresh, sliced, ½ cup	49	82
Brussels sprouts, cooked, ½ cup	48	80
Grapefruit, ½ medium	39	65
Broccoli, raw, ½ cup	39	65
Tomato juice, ¾ cup	33	55
Cantaloupe, ½ cup	29	48
Cabbage, cooked, ½ cup	28	47
Cauliflower, raw, ½ cup	26	43

### Inhibitors of absorption

There used to be some concern about tea consumption and iron absorption. However, tea consumption does not influence iron status in Western populations in which most people have adequate iron stores as determined by serum ferritin concentrations. Only in individuals with marginal iron status, may there be a negative association between tea consumption and iron status [17].

There has also been a theoretical concern about the larger intake of phytates that comes from following a plant-based diet inhibiting absorption of some minerals, such as iron. However, it turns out that the intestinal flora of vegetarians act to degrade phytate, thus allowing for good absorption of minerals. One recent study concludes that, “it was the vegetarians’ microbiota that particularly degraded up to 100% phytate to myo-inositol phosphate products [18].”

While some studies have found that oxalic acid (present in spinach, silver beet and beetroot leaves) may inhibit iron absorption, recent studies suggest that its effects are relatively insignificant [19]. Calcium has also been considered an inhibitor of both heme and non-heme iron absorption, but recent research suggests that, over a long period of time, calcium has a limited effect on iron absorption (possibly due to an adaptive physiological response) [20] Nevertheless, it may be best to avoid consuming large calcium supplements with meals.

### Clinical considerations

Vegetarians and vegans who eat a varied and well-balanced diet are not at any greater risk of iron deficiency anemia than non-vegetarians, even though iron stores of vegetarians may be lower than in non-vegetarians [3,21-23].

Iron deficiency anemia occurs in 5–12% of otherwise healthy premenopausal women [24,25] and is usually due to menstrual loss, increased demands in pregnancy and breast feeding, or dietary deficiency [26]. Blood loss from the GI tract is the most common cause in adult men and postmenopausal women [27-32].

Gastrointestinal conditions, such as celiac disease and inflammatory bowel disease, as well as chronic kidney disease, cancer, and chronic heart failure increase the risk for anemia and iron deficiency [33-41]. Underlying conditions should be treated concurrently.

The diagnostic criteria for anemia in iron deficiency anemia vary between studies: Hgb less than 10–11.5 g/dl for women and less than 12.5–13.8 g/dl for men. The lower limit of the normal range of hemoglobin concentration, for the laboratory performing the test, should therefore be used to define anemia [42].

For anemic patients, both the mean corpuscular volume (MCV) and the mean corpuscular hemoglobin concentration (MCHC) will also have values below the normal range for the laboratory performing the test. Reference range values for MCV and MCHC are 83-97 fL and 32-36 g/dL, respectively.

Serum ferritin concentration is a valuable test for iron deficiency anemia [43]. A serum ferritin concentration of less than 12  $\mu$ g/dl is diagnostic of iron deficiency anemia [44]. However, elevated ferritin levels are usually due to causes such as acute or chronic inflammation, chronic alcohol consumption, liver disease, renal failure, metabolic syndrome, or malignancy rather than iron overload [45], so serum ferritin may be above 12–15  $\mu$ g/dl in patients with iron deficiency anemia and concurrent chronic inflammation, malignancy, or hepatic disease. If the concentration is greater than 100  $\mu$ g/dl, iron deficiency anemia is almost certainly not present [44].

To avoid false negative results (high ferritin levels in spite of iron deficiency), an acute phase reaction should be excluded by taking a history and measuring the C-reactive protein or erythrocyte sedimentation rate [45].

The cornerstone of preventing iron deficiency in patients following a plant-based diet is to instruct patients to include plant foods high in iron in their diet. These foods can also be used to treat mild deficiencies while they can serve as an adjunct to iron supplementation. Patients should also be counseled to include iron absorption enhancers in the prevention and treatment of iron deficiency anemia.

Treatment of an underlying cause should prevent further iron loss, but all anemic patients should have iron supplementation

both to correct anemia and replenish body stores [46]. This is achieved affordably with ferrous sulfate 200 mg twice daily. Lower doses may be as effective and better tolerated [47,48] and should be considered in patients not tolerating traditional doses. Other iron compounds (eg, ferrous fumarate, ferrous gluconate) or formulations (iron suspensions) may also be tolerated better than ferrous sulfate. Oral iron should be continued for 3 months after the iron deficiency has been corrected so that stores are replenished [49].

Intestinal iron absorption is limited. The maximum rate of absorption of 100 mg of oral iron supplement is 20% to 25% and is reached only in the late stage of iron deficiency. Latent iron deficiency and iron deficiency anemia correspond to mean absorption rates of 10% and 13%, respectively, whereas healthy males absorb 5% and healthy females 5.6% [49,50].

Dose-dependent gastrointestinal side effects of iron supplementation hinder compliance and result in nonadherence in up to 50% of patients [51]. For those intolerant or not responding to oral iron, three parenteral preparations are available [49].

### Discussion

Patients following a plant-based diet, and their physicians, need not be concerned that a plant-based diet will lead to iron deficiency anemia any more than an omnivorous diet will. However, that doesn't mean that iron deficiency anemia shouldn't be tested for when indicated. Iron deficiency is prevalent in the general population; an average of 5.6% of the U.S. population met the criteria for anemia and 1.5% for moderate-severe anemia [52]. Menstruating and pregnant women are at particular risk [52].

The vegetarian and vegan patient should be instructed as to which plant foods are high in iron and how to enhance absorption. Plenty of foods that are high in iron can be chosen. If this is not enough and anemia results, then supplements are available and can be employed the same way as with omnivorous patients.

There are now several million Americans who follow either a vegan or vegetarian diet. Many physicians have started to prescribe a plant-based diet for the prevention and treatment of several diseases such as coronary artery disease and type 2 diabetes. By prescribing a diet that has ample iron-containing foods, along with the foods that enhance absorption, the physician can help ensure an adequate iron status in their patients following a plant-based diet.

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