

Research Article Volume 21 Issue 4 - June 2023 DOI: 10.19080/CTBEB.2023.21.556066



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Development and Validation of RP-HPLC Method for the Quantitative Analysis of Bempedoic acid in Bulk and Pharmaceutical Dosage form Using Surface Response Methodology

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Submission: June 02, 2023; Published: June 15, 2023

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Abstract

Objectives: A specific and novel RP-HPLC method has been optimized for the quantitative analysis of bempedoic acid using response surface methodology.

Method: Response surface methodology (RSM) helps in studying the empirical relationship between one or more measured responses and many independent variables in the form of a polynomial equation. Mapping of those responses related to the experimental domain helps in generating an optimized method. In this present study, the empirical relationship between measured responses like % organic phase, column temperature, and flow rate and independent variables like retention time and theoretical plates are drawn in the form of a polynomial equation and an optimized method was developed from that using Box-Behnken Design (BBD).

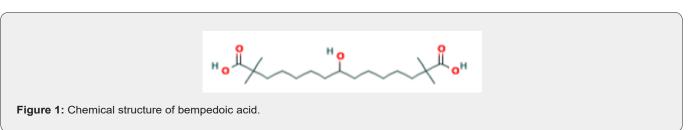
Results: The analytes chromatogram was run through SB C18 100x1.8mm, 2µm. The optimized data from Design Expert software consists of 0.01% OPA: Acetonitrile (56.05:43.95%, v/v) as mobile phase pumped through column with 1.02ml/min flow rate at 30.09°C gave highest desirable function of 1. The retention times of the drug were found to be 2.231 min. The optimized method was validated as per instructions given in ICH Q2 (R1) guidelines.

Conclusion: Based on the results of the analysis of variance, the selected model for the responses like retention time and tailing factor were found to be significant with p=0.05. 2D contour plots were inured to visualize the effect of factors and their interactions on the response. Validation of design was done using actual plots vs. predicted values for responses. All the validation parameter results were within limit.c

Introduction

Bempedoic Acid [1] is marked under the brand name Nexletol, which is a prodrug approved for the treatment of hypercholestrerolemia. Generally, this drug requires activation in the liver, the very long chain acyl-CoA Synthetase-1 enzyme helps for the activation of the drug as BETC-1002-CoA, the active metabolite of the drug. ATP synthase (properly known as ATP lyase) is responsible for the synthesis of cholesterol. The active metabolite of the drug i.e., BETC-1002-CoA directly inhibits the activity of ATP synthase which leads to up-regulation of the LDL cholesterol receptor and that reduces serum LDL-C through increased uptake and clearance of LDL in the liver. The IUPAC name of the drug is 8-hydroxy-2,214,14-tetra methyl pentadecanedoic acid and the chemical structure of the drug is shown in (Figure1). Since this drug was approved by FDA in February 2020, till this time no analytical method was reported for the estimation of bempedoic acid alone, even though the methods are available for combination of drug. Only phase studies [2] were done to know the safety and efficacy of the drug. Hence the present work was focused on the development and validation of determination of bempedoic acid by using RSM with the help of Design Expert software [3]. This approach helps in good experimental designs, risk assessment, ruggedness, and robustness testing in much more dynamic when compared with the general approach. For the assortment of initial chromatographic conditions, a 2³ factorial design [4] was selected with three factors at two levels in RSM [5]. RSM helps in studying the empirical relationship between one or more measured responses and several independent variables in the form of a polynomial equation. Mapping of those

responses related to the experimental domain helps in generating an optimized method. Optimization of the method for the present study was done with the help of the Box-Behnken design [6], which is the most popular statistical experimental design used in RSM.



Materials And Methods

Bempedoic acid is gifted from Spectrum Labs. Acetonitrile, phosphate buffer, methanol, potassium dihydrogen orthophosphate buffer, and orthophosphoric acid are purchased from Rankem. The formulation Nexletol (Bempedoic acid 180mg) was brought from the local market. Water HPLC 2695 system with photodiode array detector integrated with Empower 2 software is used for HPLC study. Design Expert® (13.0.5.0x64) modeling software (Stat-Ease Inc., Minneapolis, MN, USA) was worn for the production of 2D contour plots and 3D surface plots.

0.1%OPA Buffer: Take 1 ml of orthophosphoric acid and diluted it to 1000 ml with HPLC grade water.

Preparation of Mobile Phase: The mobile phase was ready by adding HPLC grade acetonitrile and 0.1% OPA in the ratio of 50:50.

Preparation of Diluent: HPLC grade acetonitrile and water in the ratio of 50:50 is used as diluent.

Preparation of Standard Stock Solutions: The standard stock solution of bempedoic acid was prepared by accurately weighing 180mg of bempedoic acid and transferred to 200ml volumetric flasks and add 3/4th of diluents to this flask and sonicate the solution for 10 minutes. Final volume was made up of diluents. One milliliter of the above prepared solution was

transferred to a 10-ml volumetric flask, and then final volume was made with a diluent (standard solution). The stock solution was diluted as per the requirement.

Preparation of Sample Solution: Five tablets (Nexletol) were weighed and crushed. Quantity of powder equivalent to 180mg of bempedoic acid was taken in a 200ml volumetric flask, and $3/4^{\text{th}}$ of diluents was added to this flask, sonicated it for 10 min and the final volume was made up with diluent. The prepared solution was filtered through a 0.45µm membrane filter and further diluted as per requirement.

Optimized Chromatographic Conditions: The initial trials are required to optimize the final method. Chromatographic separation was accomplished on SB C18 100x1.8mm, 2µm column at 30.09°C. A mixture 0.01% OPA: Acetonitrile (56.05:43.95%, v/v) was used a mobile phase with a flow rate of 1.02ml/min. The detection of chromatogram was done at 220nm of UV.

Experimental Design: The method was optimized using Box-Behnken design. Total 3 factors i.e., % organic content, temperature of the column and flow rate were optimized. Hence, Box-Behnken design was used to optimize these parameters at three levels (high, mid, and low). Different ranges of three parameters 40-60% acetonitrile, column temperature 27-33°C and flow rate of 0.9-1.1ml/min were considered as shown in (Table 1).

Table 1: Design summary of Box-Behnken Design.

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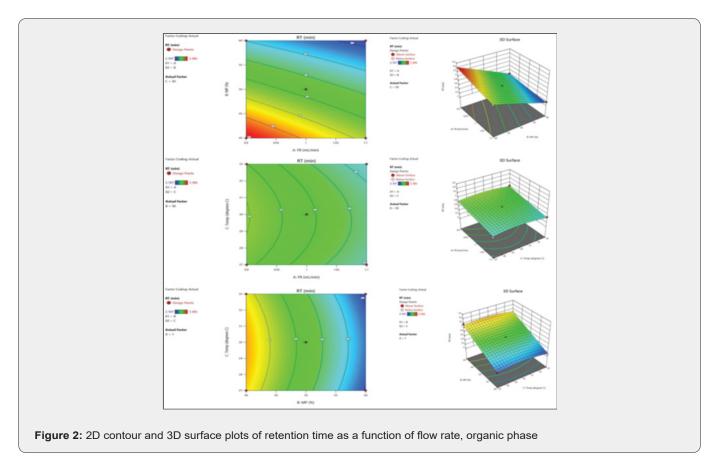
File version: 13.0.12.0							
	Study type: Response Surface						
	Design type: Box-Behnken design						
CQA: Retention time, Theoretical plates Runs: 17							
CMPs	Units	Туре	Subtype	Min.	Max.		
% Organic Composition	%v/v	Numeric	Continuous	40	60		
Flow Rate	ml/min	Numeric	Continuous	0.9	1.1		
Column Temperature	°C	Numeric	Continuous	27	33		

Method Validation: The validation of the optimized analytical method was done as per the guidelines of International Conference on Harmonization Q2(R1) [7].

Linearity: Standard calibration curve was prepared with five different concentrations over the range of $22.5-112.5\mu g/m$ l. Linear calibration curve was generated between peak area and drug concentration. The linearity was examined using linear

regression, which was calculated by the least square regression method shown in the (Figure 8).

Accuracy: Accuracy was performed by adding known amount of sample to the 0.5ml, 1.0ml and 1.5ml of 900µg/ml standard solution of the drug which gives 50%, 100%, and 150% levels. The assessment is performed in triplicate by the optimized method. Percentage recovery was measured.



Precision: Precision of the optimized method was measured by studying the system, method, and intermediate precision. Six 67.5μ g/ml standard solutions of pharmaceutical formulation were injected on the same day and next day of the preparation of samples and the % RSD of the peak area was calculated.

Specificity: The specificity was deliberate by injecting the blank, placebo, and pharmaceutical preparation of drug, several times on several days. It was revealed that there was no interference of peak in the region of Bempadoic Acid in chromatogram for the blank, and placebo. Representative chromatograms of blank, placebo, and drug standard are shown in (Figures 9&10).

Limits of detection (LOD) and limits of quantitation (LOQ): LOD and LOQ values were calculated from the signal-tonoise ratio method and the chromatograms are shown in (Figure 11&12).

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Robustness: To verify the method effectiveness when minor changes occurred in optimized method parameters made such as proportion of organic concentration in the mobile phase (40-60%), flow rate (0.9-1.1ml/min), and temperature of the column (25-35°C). %RSD of the mentioned conditions was calculated.

System suitability: The system suitability was measured by taking six replicates of the drug at same concentration i.e., $67.5\mu g/m$ l. The acceptance criteria were ± 2% for all the components such as the percent coefficient of variation (% CV) for the peak area, retention time of drug, USP theoretical plate number, and tailing factor.

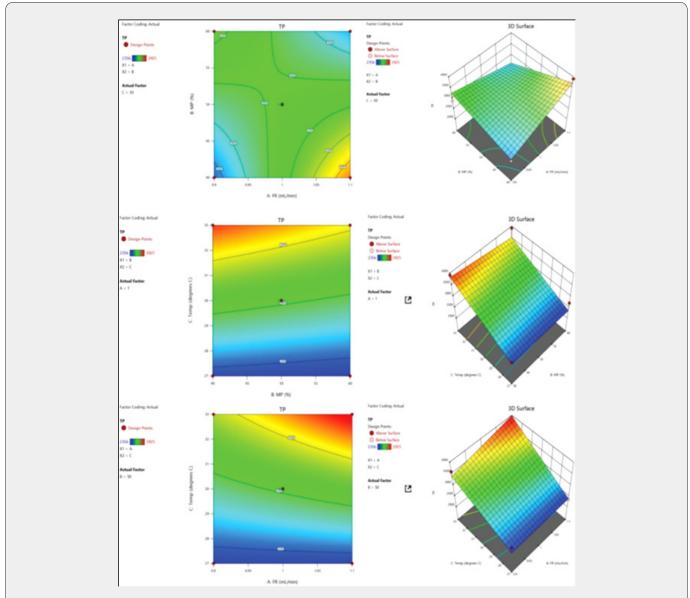
Forced degradation studies [8]

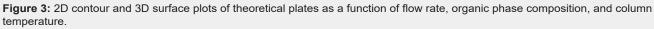
Acid hydrolysis: To 1ml of stock solution of the pharmaceutical formulation add 1ml of 2N HCl. The resulting solution was placed

in radley apparatus for reflux with constant stirring at 70°C for 60 min. The refluxed solution was neutralized with 2N NaOH and diluted up to 10ml with mobile phase.

pharmaceutical formulation add 1ml of 2N NaOH solution. The degradation solution was placed in radley apparatus for reflux with constant stirring at 70°C for 60 min. The refluxed solution was neutralized with 2N HCl and diluted up to 10ml with mobile phase.

Base hydrolysis: To 1ml of stock solution of the





Neutral hydrolysis: Dilute 1ml of stock solution of the pharmaceutical formulation to 10ml with HPLC grade water. The degradation solution was placed in radley apparatus for reflux with constant stirring at 70°C for 4h.

Oxidative study: To 1ml of stock solution of the pharmaceutical formulation add 1ml of 20% H_2O_2 solution. Then the degradation sample was set aside in dark area without interruption at room

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temperature for 4h and dilutes the solution up to 10ml with mobile phase.

Thermal degradation: The powdered form of pharmaceutical formulation was place in hot air oven at 70°C for 60 min. The dilution of the pharmaceutical formulation was done with mobile phase and analyzed that using the HPLC system.

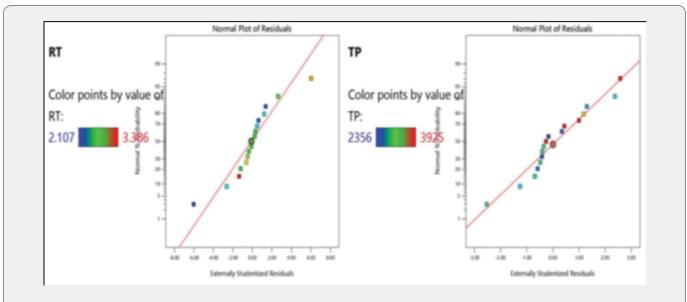


Figure 4: Normal plot of studentized residuals for retention time and theoretical plates.

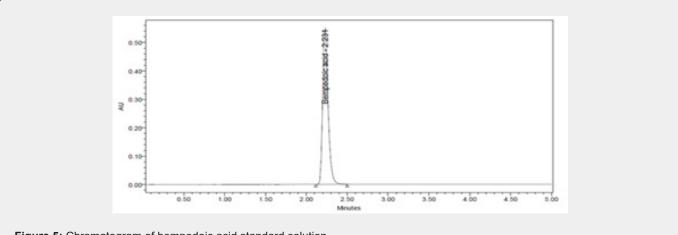
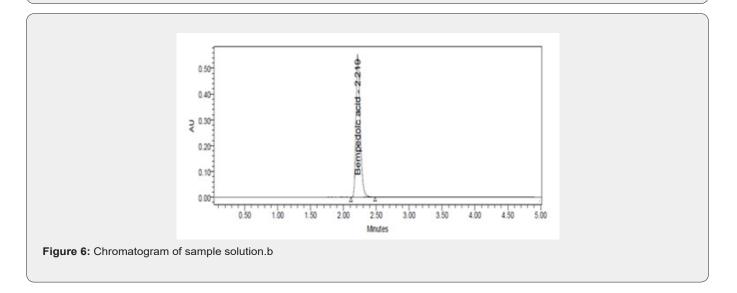


Figure 5: Chromatogram of bempedoic acid standard solution.



Run	S. No.	% Organic Composition	Flow Rate	Column Temperature	Retention Time	Theoretical Plates
17	1	50	1	30	2.223	5924
11	2	50	0.9	33	2.843	5610
16	3	50	1	30	2.723	5924
8	4	60	1	33	2.107	4916
6	5	60	1	27	2.197	5714
9	6	50	0.9	27	2.803	5614
4	7	60	1.1	30	2.172	5524
15	8	50	1	30	2.723	5924
2	9	60	0.9	30	2.405	5894
12	10	50	1.1	33	2.459	5925
5	11	40	1	27	3.161	5482
1	12	40	0.9	30	3.386	5441
14	13	50	1	30	2.313	5849
13	14	50	1	30	2.713	5849
3	15	40	1.1	30	2.346	5911
7	16	40	1	33	3.045	5916
10	17	50	1.1	27	2.451	5356

Table 2: Box-Behnken experimental design with responses.

Table 3: Created ANOVA table using Box-Behnken design for retention time.

	Analysis of variance for the quadratic mode of the response surface						
		Analysis of variance (T	ype-III of the partia	al sum of squares)			
Source	Sum of Squares	Degree of Freedom	Mean Square	p-value	F value	Inference	
Model	1.969	9	0.2192	<0.0001	130.98	Significant	
A-FR	0.2483	1	0.2481	< 0.0001	148.36	Significant	
B-MP	1.669	1	1.669	<0.0001	999.52	Significant	
C-Temp	0.0032	1	0.0032	0.2142	1.88		
AB	0.0109	1	0.0109	0.0392	6.39	Significant	
AC	0.0004	1	0.0004	0.7073	0.1532		
BC	0.0003	1	0.0003	0.7599	0.1011		
A ²	0.0002	1	0.0002	0.7767	0.0868		
B^2	0.0001	1	0.0001	0.7859	0.0798		
C ²	0.0372	1	0.0372	0.0022	22.189	Significant	
Residual	0.0118	7	0.0118				

Table 4: Fit statistics.

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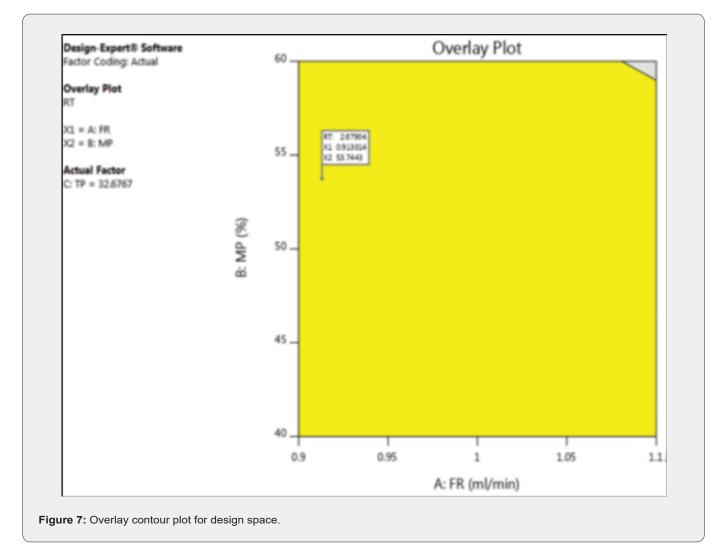
Std. Dev.	0.04089	R ²	0.994
Mean	2.69	Adjusted R ²	0.9869
C.V. %	1.53	Predicted R ²	0.9089
Adequate precision			60.55

	Analysis of variance for the response surface quadratic model							
	Analysis of variance table (Type III of Partial sum of squares)							
Source	Sum of squares	degree of freedom	Mean square	F value	p-value	Inference		
	Model 4.55E+06 6		7.59E+05	7.03	0.0009	Significant		
A-FR	1.67E+05	1	1.67E+05	2.25	0.1656			
B-MP	61600.52	1	61601.5	0.8233	0.3855			
C-Temp	3.38E+06	1	3.38E+06	45.21	< 0.0001	Significant		
AB	8.46E+05	1	8.47E+05	11.3	0.0072	Significant		
AC	82081.25	1	82082.26	1.11	0.3195			
BC	13456.01	1	13456.01	0.1798	0.6805			
Residual	7.48E+05	10	74812.04					

Table 5: ANOVA table for theoretical plates using Box-Behnken design.

Photo degradation: The powdered form of pharmaceutical formulation was evenly spread in a petri dish and was exposed to UV light with NLT 2000 lux power intensity for 24h. The dilution

of the powdered form of pharmaceutical formulation was done with mobile phase and analyzed that using the HPLC system.



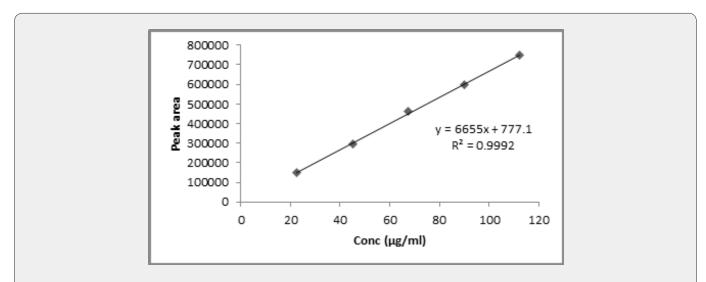
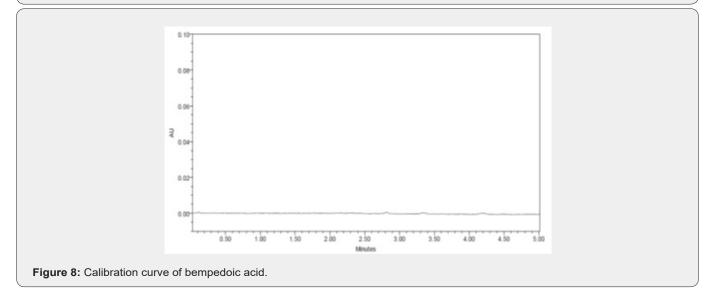


Figure 8: Calibration curve of bempedoic acid.



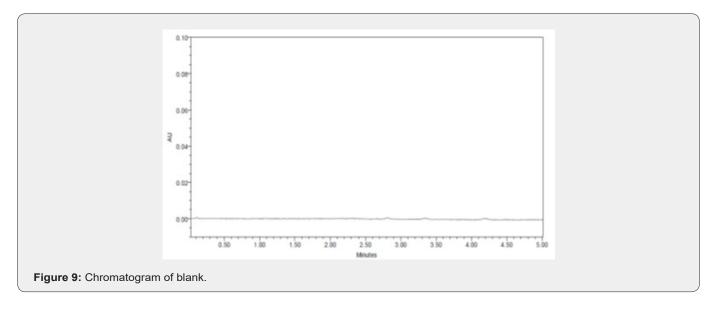


Table 6: Fit statistics.

Std. Dev.	273.52	R ²	0.9589
Mean	5816	Adjusted R ²	0.9012
C.V. %	9.67	Predicted R ²	0.9275
Adequate precision			9.67

 Table 7: Final optimized HPLC chromatographic conditions.

Chromatographic condition	Value	
Mobile phase	0.1 % OPA (46.3%): Acetonitrile (53.7%)	
Flow rate	1.02 ml/min	
Column temperature	30.09 °C	

Table 8: Responses of the optimized method.

S. No.	Response variables	Predicted value	Actual value	Desirable range
1	Retention time (min)	2.379	2.231	2.19436-2.46364
2	Theoretical plates	5921.66	5621	5203.18-6423

Table 9: Results of the validation parameters.

S. No.	Parameters	Results		
	Linearit	у		
	Linearity range (µg/ml)	22.5-112.5		
1	Correlation coefficient	0.999		
	Regression equation	y = 6655x + 777.1		
2	Accuracy (% reco	very) (n=3)		
2	50%, 100%, 150% levels	Between 99.3 and 100.5		
	Precision (% RSD of p	eak area) (n=6)		
2	System precision	1.2		
3	Repeatability	0.91		
	Intermediate precision	1.12		
	Sensitivity			
4	LOD (µg/ml)	0.77		
	LOQ (µg/ml)	2.34		
	Robustness (% RSD of peak area)			
	Flow rate (±0.1 ml/min)	1.15		
	Organic phase (±10%)	0.7		
	Temperature (±5°C)	0.85		
	Robustness (% RSD of tailing factor)			
	Flow rate (±0.1 ml/min)	0.897		
5	Organic phase (±10%)	1.51		
	Temperature (±5°C)	0.78		
	Robustness (% RSD of numbe	er of theoretical plates)		
	Flow rate (±0.1 ml/min)	0.879		
	Organic phase (±10%)	1.24		
	Temperature (±5°C)	0.69		

	System sui	tability
6	Retention time (min)	2.231
0	Tailing factor	1.27
	Theoretical plate number	5621

Results

Statistical analysis of experimental data by designexpert software

ANOVA was used to study the significance level of the model.

The Model F-value of 130.98 implies the model is significant for the responses retention time and theoretical plates given in the (Tables 3-5). There is only a 0.01% possibility that an F-value this large could occur due to noise. p<0.0500 shows that model terms are significant. In this case, A, B, AB and C² are significant model terms.

The predicted R^2 of 0.9089 is in reasonable harmony with the adjusted R^2 of 0.9869; that is, the variation is less than 0.2. Adequate precision determines the signal-to-noise ratio. A ratio more than 4 is desirable. S/N ratio of 60.55 indicates an adequate signal presented in the (Table 4). This model can be helped to navigate the design space. 2D contour and 3D surface plots were studied to visualize the effect of factors and their effects on the responses using the Design Expert® software. The region in dark blue represents lower values and with dark red represents higher values. The regions in light blue, green, and yellow represent intermediary values.

By considering the above 2D Contour and 3D surface plots of retention time shown in (Figure 2), it was established that at a higher flow rate, higher temperature, and higher organic phase composition lower will be the value of retention time.

The Model F-value of 7.03 implies the model is significant. There is only a 0.09% chance that an F-value this large could occur due to noise. p<0.0500 indicates that model terms are significant. In this case, C and AB are significant model terms.

The predicted R^2 of 0.9275 is in sensible agreement with the adjusted R^2 of 0.9012; that is, the variation is less than 0.2. Adequate precision determines the signal-to-noise ratio. A ratio more than 4 is desirable. S/N ratio of 9.67 indicates an adequate signal. This model can be used to plot a route to the design space given in the (Table 6).

By considering the above 2D contour and 3D surface plots of theoretical plates shown in (Figure 3), it was found that at a higher temperature, higher flow rate, and lower the organic phase composition higher will be the value of theoretical plates.

Design validation

From the normal plot of studentized residuals for the two responses represented in (Figure 4), it was observed that the selected models for the respective responses were fit for the selected design as these plots represents straight line. It was further concluded from the (Tables 3&4) that the selected models were significant with p<0.05. Hence, the selected models were fit for the design engaged in this work.

Optimization by desirability function

Desirability was applied to get an optimum set of conditions based on the particular goals and limitations for each response. This desirability function relays on a scale of desirability function ranges between d = 0 for a totally undesirable response, to d = 1 for a completely desirable response. Based on the particular goals and limitations for the retention time (minimum) and theoretical plates (maximum) a composite desirability (D) was obtained at 1. To validate these optimum set of conditions given in the (Table 7), three replicate injections of drug were analyzed to measure if their experimental retention time and theoretical plates were within the predicted ranges shown in the (Table 8) and the corresponding optimized standard and sample chromatograms were shown in the (Figures 5 & 6) respectively.

Over lay plot

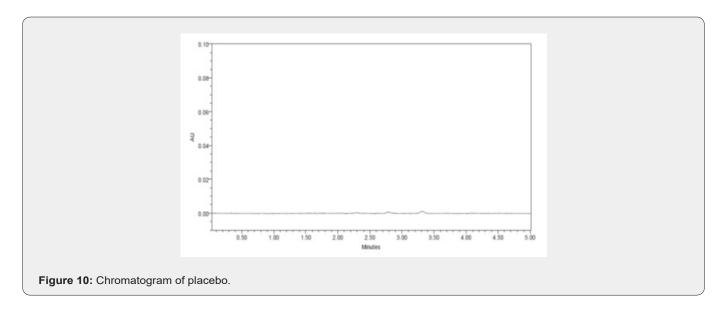
The overlay counter plot presents the QbD design space where the method obeys the mean performance goals and robustness criteria shown in (Figure 7). The flag represents optimized permutation of the three selected independent factors, which helps to set desirability of minimum retention time and maximum theoretical plates.

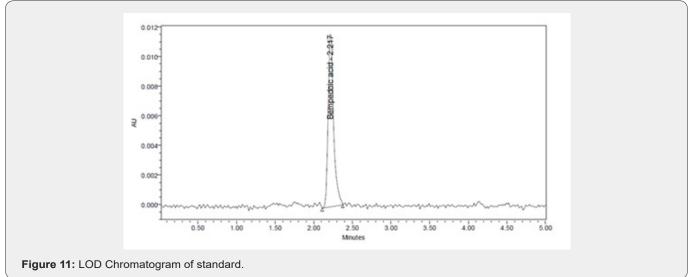
Method validation

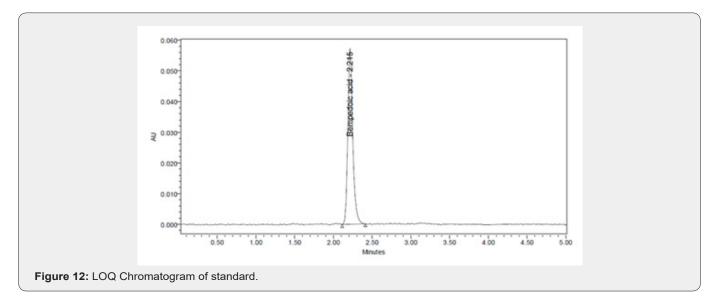
The method was linear over the concentration range of 22.5-112.5µg/ml with correlation coefficient of 0.999. For the accuracy studies was performed at 50, 100, and 150% levels and the % drug recovery was noted to be within 99.3-100.5%. Intermediate precision and repeatability were performed, and the % RSD values were less than 2%. LOD and LOQ values were found to be 0.77µg/ml and 2.34µg/ml. Robustness of the method was done by making slight changes in the experimental conditions such as % organic composition, flow rate, and temperature and % RSD values were observed at less than 2%. The summary data of the method validation parameters is shown in (Table 9).

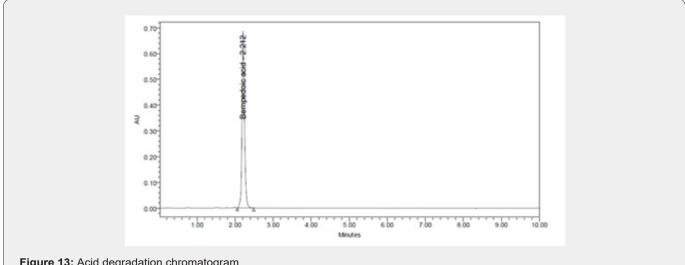
Forced degradation studies

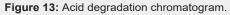
Forced degradation studies were done at various conditions such as acidic, basic, peroxide, thermal, photolytic, and hydrolytic. Results of forced degradation studies are given in Table 10 and the chromatograms are shown in (Figures 13-18).

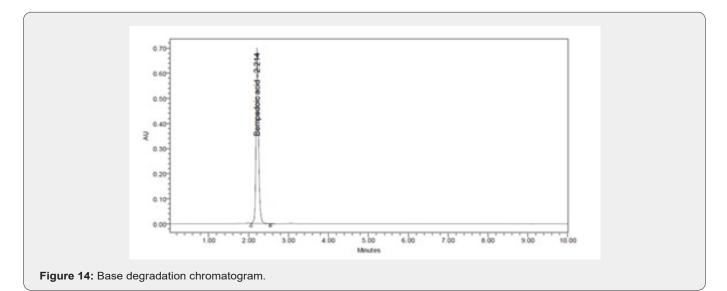


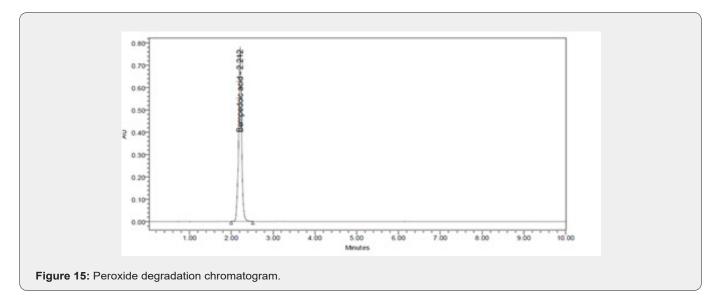


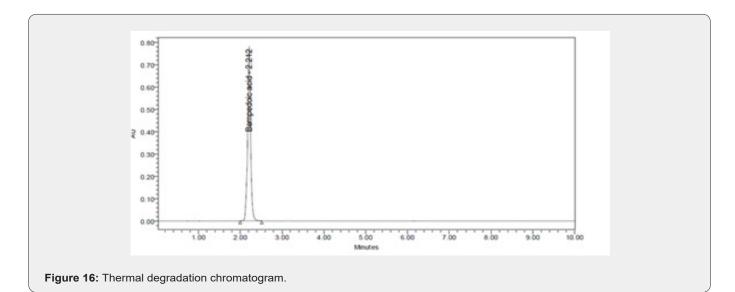


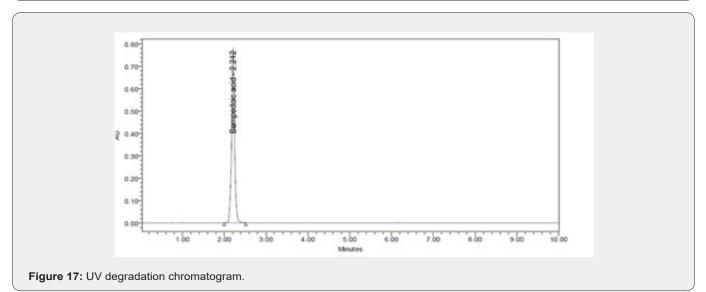


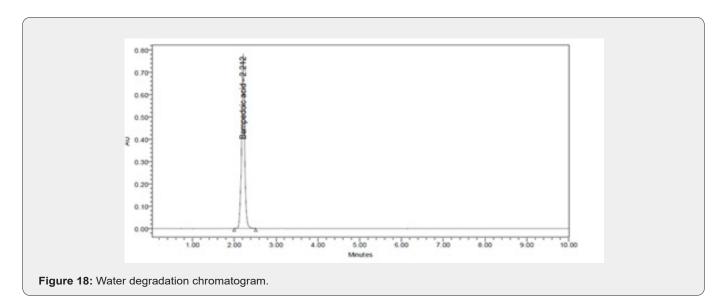












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Discussion

A simple, sensitive, robust, accurate, and precise RP-HPLC method was developed for the determination of bempedoic acid using the Response Surface Methodology. The % of organic content in the mobile phase, Column temperature, and flow rate were selected as Critical Method Parameters for Critical Quality Attributes i.e., retention time and theoretical plates. The Critical Method Parameters were analytically optimized using the Box-Behnken design. Mobile phase 0.1 % OPA (46.3%): Acetonitrile (53.7%), pumped at a flow rate of 1.02ml / min is finalized as optimized chromatographic conditions. The significant factors impacting each response were identified using 2D contour and 3D surface plot. The ANOVA test was applied to acquire the p value, R², and the equations for each response by including only significant terms. Utilization of RSM delivers a better awareness of method development. The retention time of the drug for the

developed method was found to be 2.231 min. Theoretical plates and tailing factor was found to be within the limits. The validation of the developed method was done as per the ICH Q2 (R1) guidelines Degradation studies for the developed method were performed in various stress conditions, which was started with a shorter duration of exposure (in case of acidic, basic, and oxidative stress), but the drug did not exhibit much deterioration; therefore, the duration of the exposure was extended to produce a significant level of degradation. Here, comparatively more degradation was observed under acidic conditions due to hydrolysis of the drug. Under long-wavelength UV exposure (photolysis 24 h) and moist heat at 70 °C (thermal degradation 24 h), the drug has shown less degradation. Our approach has shown better quantification of the drug in different stress conditions due to the improved peak shape and absence of interference at the retention time of the drug. The Bempadoic Acid showed minimum degradation in all stress conditions except with acid-induced hydrolysis.

Table 10: Results of forced degradation studies.

Drug	Degradation Condition	% Recovery	% Drug Degraded
	Acid	93.81	6.19
	Alkali	95.42	4.58
Bempedoic Acid	Oxidation	95.76	4.24
	Thermal	97.15	2.85
	UV	98.31	1.69
	Water	99.37	0.63

Conclusion

Based on the results of the Analysis of variance, the selected model for the responses like retention time and tailing factor were found to be significant with p=0.05. 2D contour plots were inured to visualize the effect of factors and their interactions on the response. Validation of design was done using actual plots vs. predicted values for responses. All the validation parameter results were within limit.

Acknowledgement

The authors are grateful to Spectrum Pharma Research Solutions., Hyderabad for providing gift samples, and the authors are also obliged to V. V. Institute of Pharmaceutical Science, Gudlavalleru for providing the necessary facilities to carry out the research work.

References

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- 1. Saeed A, Ballantyne CM (2018) Bempedoic Acid (ETC-1002): A Current Review. Clinical Cardiology 36(2): 257-264.
- 2. Ray KK, Bays HE, Catapano AL, Lalwani ND, Bloedon LT, et al. (2019)

Safety and Efficacy of Bempedoic Acid to reduce LDL Cholesterol. New England Journal of Medicine 280(11): 1022-1032.

- Fukuda IM, Pinto CFF, Moreira CS, Saviano AM, Lourenco FR (2018) Design of experiments (DoE) applied to pharmaceutical and analytical quality by design (QbD). Brazilian Journal of Pharmaceutical Sciences 54:e01006.
- 4. Dipen G, Saurabh CK (2017) A Review on Analytical Quality by Design. International Journal of Pharmaceutical Sciences Review and Research 44(2): 96-102.
- Sahu PK, Nageswara RR, Teresa C, Suryakanta S, Chandra Sekhar P, et al. (2018) An overview of experimental designs in HPLC method development and validation. Journal of Pharmaceutical and Biomedical Analysis 147: 590-611.
- Candioti LV, de Zan MM, Camara MS, Goicoechea HC (2014) Experimental design and multiple response optimization. Using the desirability function in analytical methods development. Talanta 124: 123-38.
- ICH Harmonised Tripartite Guideline: Validation of Analytical Procedures: Text and Methodology Q2 (R1), Current STEP 4 Version. Geneva: International Conference on Harmonisation; 2005.
- Blessy M, Patel DR, Prajapati NP, Agarwal YK (2014) Development of forced degradation and stability indicating studies of drugs-a review. Journal of Pharmaceutical Analysis 4(3): 159-65.



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